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a **single replacement** reaction is a reaction in which one element takes the place of another element in a compound. In a **single replacement** reaction, or **single** ...

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Practice Problem **Answers**. Back to **Single Replacement**. Write correct formulas for the products in these **single replacement** reactions. Unbalanced **Answers**

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During **single replacement**, ... Your teacher may require this, but the ChemTeam will only provide some of the following **answers** balanced. The rest are up to you!!

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In a **single replacement** reaction, the element replaces the one of the ions in the compound, depending whether the element is more likely to give away electrons or ...

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**single-replacement** reaction (  $\text{single replacement reaction}$  ) ( chemistry ) A chemical reaction in which an element repla

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A **single replacement** reaction occurs when two different cations switch places to combine with the same anion. One element forms a compound while another element ...

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[answers.yahoo.com/question/index?qid=20081013124601AA131qr](#)

Resolved · Last updated: Oct 13, 2008 · 2 posts · First post: Oct 13, 2008

Oct 13, 2008 · Best **Answer**: **Single replacement** reactions will take one of two forms.  
 $M + AX \rightarrow MX + A$  or  $MX + Y \rightarrow MY + X$  These reactions typically involve oxidation ...

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Resolved · 1 total answer · Published Jan 10, 2008

Jan 10, 2008 · Best **Answer**: You figured it out... the last piece is absolutely right. It's  $KOH + Li \rightarrow LiOH + K$  In the earlier part of your thinking, you were trying to ...

### [Single-Replacement and Double-Replacement Reactions ...](#)

<https://www.teachervision.com/chemical-mixtures/printable/57628.html>

**Help** students visualize **single-replacement** and double-replacement reactions with these chemistry diagrams. This science printable can be used as a class handout or ...

### [Worksheet #4: Single-Replacement Reactions Step 1 - Write ...](#)

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Worksheet #4: **Single-Replacement** Reactions Step 1 - Write the formulas of the reactants on the left of the yield sign Step 2 - Look at the Activity Series on page 333 ...

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